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Chemical and
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(Classification of
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ozone formation,

chlorofluorocarbons,

CFC ' s 1.2 Chemistry

and Matter Chemistry

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Chapter 1 Section 2

MATTER and Its

Properties Matter –

Anything that has

MASS and VOLUME.

ATOMS and

MOLECULES the basic

building blocks of

matter. ATOM-- the

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smallest unit of an element that still maintains the properties that element. ELEMENT— a pure substance made up of only one type of atom.

MATTER and CHANGE
Chapter 1 Section 1
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about atomic radius

and ionic radius.

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choice that best

completes the

statement or answers

the question. 1.

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and ChangesMatter

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Section 3.1 Properties
of Matter pages

70–75 Problem-

Solving Lab 1. Explain

why the flow of a

compressed gas must

be controlled for

practical and safe

use.

Matter—Properties

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and Changes Matter

— Properties and
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Periodic Law Section
6.1 Development of
the Modern Periodic
Table 1. octaves 2.
eight 3. nine 4.
accepted 5. Dmitri
Mendeleev 6. atomic
mass 7. elements 8.
atomic number 9.
Henry Moseley 10.
protons 11. periodic
law 12. properties 13.
14 ...

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Chapter 5 25

Electrons in Atoms

Reviewing

Vocabulary Match the

definition in Column

A with the term in

Column B. Column A

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Column B 1. The set of frequencies of the electromagnetic waves emitted by the atoms of an element

2. The minimum amount of energy that can be lost or gained by ...

Chapter Assessment
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extensive property.

intensive property.

extensive properties

(examples) intensive

property (examples)

a property that

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depends on the amount of matter in a sample. a property that depends on the type of matter in a sample not.... mass, volume, length, weight.

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Chapter 11 31

number of moles of

known substance C H
C H C H O O O CO CO
CO H O H O H O

number of moles of
unknown substance
O CO H O C H CO H O
C H O H O C H O CO

use mole ratio 5 mol
O 2 1 mol C 3 H 8 3

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mol CO₂ 1 mol C₃H₈

4 mol H₂O 1 mol C₃H₈

5 mol O₂

mol O₂

Answers

MATH SKILLS

TRANSPARENCY

MASTER 16

A “chemical change” is a change that produces matter with a different composition than the original matter.

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A chemical change is a change in which one or more substances are converted into different substances.

Heat and light are often evidence of a chemical change.

Properties of
Compounds Quite
different properties
than their
component
elements.

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