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1,2,3 Limiting Reactant

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GabbyLuo. Terms in this set (11)

Stoichiometry. The portion of chemistry dealing with numerical relationships in chemical reactions; the calculation of quantities of substances involved in chemical equations. Mole ratio . A conversion factor derived from the

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Stoichiometry is the calculation of the amount of substances in a chemical reaction from the balanced equation. The sample problem below is another stoichiometry problem involving ingredients of the ideal ham sandwich.

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Example 12.1.1 Kim looks in the refrigerator and finds that she has 8 slices of ham.

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how many molecules of water can I make?

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Section 12.1 The Arithmetic of Equations

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## Using Balanced Chemical Equations

Chemists use balanced chemical equations as a basis to calculate how much reactant is needed or product is formed in a reaction. Stoichiometry = calculation of quantities in chemical reactions is ...

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An expanded version of the flowchart for stoichiometric calculations illustrated in Figure 11.4.1 is shown in Figure 12.2.1. We can use the balanced chemical



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equation for the reaction and either the masses of solid reactants and products or the volumes of solutions of reactants and products to determine the amounts of other species, as illustrated in Example 7, Example 8, and Example 9.

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